

北京大学量子材料科学中心

International Center for Quantum Materials, PKU

Seminar

New Materials and Catalysts for Solar Fuels: p-Type Semiconductors Photocathodes

Andrew B. Bocarsly

Department of Chemistry, Frick Laboratory, Princeton University

Time: 4:00 pm, April.10, 2014 (Thursday)
时间: 2014年4月10日 (周四) 下午4::00
Venue: Conference Room A (607), No. 5 Science Building
地点: 理科五号楼607会议室

Abstract

Presently the conversion of optical energy to stored chemical energy focuses on two chemical transformations: the splitting of water to hydrogen and the reduction of carbon dioxide to C_1 organic products. Both of these reactions are kinetically challenging, and tend to compete with solid-state photochemical reactions that decompose the cathode materials when a photoelectrochemical cell is employed to carry out the transformation. Thus, materials that behave as stable photocathodes are needed. Similarly, electrocatalysts that enhance the multielectron reduction of H_2O or CO_2 are also desirable.¹ To these ends, we have considered both III-V materials and metal oxides having a delafossite structure as photocathodes for the reduction of CO_2 .²⁻³ These materials are found to provide a reasonable level of stability and good efficiency for the conversion of CO_2 to either formic acid or methanol. The reactivity of III-V electrodes is strongly enhanced by the use of an aromatic amine co-catalyst. We also find that CuFeO₂ is an excellent cathode material for the reduction of CO_2 to formate,⁴ while the related CuRhO₂ is an excellent material for splitting water.⁵ This latter material is found to be self-healing in the presence of oxygen providing for a superior photocathode response.

About the Speaker

Education:

Massachusetts Institute of Technology - PhD - June 1980 Physical Inorganic Chemistry Major Field: Thesis Advisor: Professor Mark Wrighton Thesis Title: "Characterization and Manipulation of Charge Transfer Processes at the Semiconductor Electrolyte Interface" University of California at Los Angeles, B.S., Magna Cum Laude with Departmental Honors in Chemistry - June 1976 Major Field: Double Major in Chemistry and Physics Research Advisor: Professor John Gladysz Research Area: "Metal Atom Cocondensations for Organometallic Synthesis" **Present Research Interests:** Elevated Temperature Proton Exchange Membrane Fuel Cells Non-hydrogen Based Fuel Cell Catalysts Photoelectrochemical and Electrochemical Conversion of CO2 to Alcohols The Photochemistry of Multielectron Intervalent Charge Transfer Complexes Cyanogels: Hydrogel Coordination Polymers as Materials Precursors

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